

A New Flavanone from *Cinnamomum Subavenium*

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ABSTRACT

One new flavanone, 6,7,4'-trihydroxy-3'-methoxyflavanone (**1**), was isolated from barks of *Cinnamomum subavenium* Miq (Lauraceae). The structure of **1** was characterized and identified by spectral analysis.

Keywords

Cinnamomum subavenium, Lauraceae, Flavanone.

Introduction

Cinnamomum subavenium Miq (Lauraceae) is a medium-sized evergreen tree, found in central to southern mainland China, Burma, Cambodia, Taiwan, Malaysia and Indonesia [1]. In the course of screening for biologically and chemically novel agents from Formosan Lauraceous plants [2-13], *C. subavenium* was chosen for further phytochemical investigation. They have identified a novel cytotoxic monoterpene, subamone, a novel sesquiterpene, subamol, five new butanolides, subamolide A-E, two new secobutanolides, secosubamolide and secosubamolide A, one new diphenyl ether, 2,2',7a,7a',7b,7b'- hexamethyldiphenyl ether, along with 47 known compounds from the stems, roots and leaves of *C. subavenium* [14-20]. In the course of screening for biologically and chemically novel agents from Formosan plants in the *Cinnamomum* species, *C. subavenium* was chosen for further phytochemical investigation. A new flavanone, 6,7,4'-trihydroxy-3'-methoxyflavanone (**1**) was isolated and identified from these barks. In this paper, we report the isolation and structural elucidation of this new compound.

6,7,4'-Trihydroxy-3'-methoxyflavanone (**1**) was obtained as a yellow amorphous powder and its molecular formula was deduced as C₁₆H₁₄O₆ by HRESIMS (*m/z* 325.0689 [M + Na]⁺; calc. 325.0688). The UV spectrum showed a λ_{\max} at 242 and 284 nm typical for a flavanone skeleton [21]. The IR spectrum exhibited strong absorption bands at 3400, 1680 and 1515 cm⁻¹ due to the hydroxyl, α,β -unsaturated ketone and aromatic C=C functionalities, respectively, in the molecule. ¹H and ¹³C NMR spectra showed the presence of three hydroxyl and one methoxy groups on the flavanone skeleton. The aromatic region of its ¹H NMR spectrum showed two singlet protons at δ 7.24 and 6.51, two *ortho*-coupled protons at δ 7.02 and 6.89 and a *meta*-coupled proton at δ 7.22. The two singlet signals were assigned to H-5 and H-8 respectively. This suggested that ring A was substituted at C-6 and C-7. In the proton NMR, a pair of double-doubles at δ 2.70 (1H, dd, J = 16.8, 3.0 Hz) and 3.06 (1H, dd, J = 16.8, 13.0 Hz) due to H-3a/H-3b and an oxy-methine proton signal at δ 5.37 (1H, dd, J = 13.0, 3.0 Hz, H-2) further attested the flavanone skeleton [21]. Significant correlations between OMe-3', H-2' H-6', and H-5', as well as H-2, and H-3, were observed in the NOESY spectrum. Therefore, the methoxy group should be located on the B-ring. The *ortho*-coupled shielded proton resonating at δ 6.87 (d, J = 8.0 Hz) was assigned at H-5' while the other proton signal at δ

7.02 (dd, $J = 8.0, 2.0$ Hz) was assigned at H-6' with the help of COSY and HMBC correlations. Further, the *meta*-coupled proton signal at δ 7.21 was assigned at H-2'. The ^{13}C NMR spectrum of **1** displayed characteristic signals of a flavanone skeleton at δ 190.5 due to an α,β -unsaturated ketone, a methylene carbon resonance at δ 44.7 (C-3) and an oxymethine carbon signal at δ 79.5 (C-2) [21]. This conclusion was reinforced by the peak correlating signals at 2.70/3.06 and 5.37 ppm observed in the ^1H - ^1H COSY spectrum. In the long-range HETCOR spectrum, H-3 (δ 2.70/3.06) shows 2J correlation to C-2 (δ 80.9) and C-4 (δ 190.5) and 3J correlation to C-1' (δ 131.2) and C-10 (δ 113.5). The methane proton H-2 (δ 5.37) shows 2J correlation to C-1' (δ 131.2) and C-3 (δ 44.7) and 3J correlation to C-2' (δ 112.9), C-6' (δ 120.2) and C-4 (δ 190.5). Elucidation of the absolute configuration at C-2 was based on the values of the coupling constants with the methylenic protons H-3 α,β ($J_{\text{ax-ax}} = 13.0$ and $J_{\text{ax-eq}} = 3.0$ Hz). The close similarity of the H-2 chemical shifts with those of the literature thus confirmed the *S*-configuration of C-2 [22,23]. Thus the structure of this compound was determined to be 6,7,4'-trihydroxy-3'-methoxyflavanone (**1**).

Experimental

General: IR, Hitachi 260-30 spectrophotometer; 1D and 2D NMR, Varian (Unity Plus) NMR spectrometer (400 MHz for ^1H NMR, 100 MHz for ^{13}C NMR); Low-resolution ESI-MS, API 3000 (Applied Biosystems); High-resolution ESI-MS, Bruker Daltonics APEX II 30e spectrometer; Silica gel 60 for CC and precoated silica gel plates (Merck) were used for TLC, visualized with 10% H_2SO_4 .

Plant material: The specimen of *C. subavenium* was collected from Taipei, Taiwan, February 2013. A voucher specimen was identified by Dr. Fu-Yuan Lu (Department of Forestry and Natural Resources College of Agriculture, National Chiayi University) and was deposited in the School of Medical and Health Sciences, Fooyin University, Kaohsiung, Taiwan.

Extraction and isolation: The barks (2.6 kg) of *C. subavenium* were extracted repeatedly with CH_2Cl_2 at room temperature for 24-48 hrs. The CH_2Cl_2 extract was dried and evaporated to root a viscous residue (34.1 g). The residue was placed on a silica gel column and eluted with CH_2Cl_2 gradually enriched with MeOH to afford 8 fractions. Fraction 7 (2.3 g) was purified by silica gel chromatography (CH_2Cl_2 -MeOH, 20:1) to give 6,7,4'-trihydroxy-3'-methoxyflavanone (**1**) (6 mg).

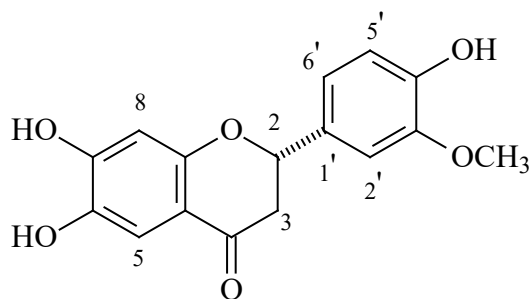


Figure 1: Structure of 6,7,4'-trihydroxy-3'-methoxyflavanone (**1**).

6,7,4'-Trihydroxy-3'-methoxyflavanone (1): Yellow amorphous powder; $[\alpha]_D^{24} = 22.8^\circ$ ($c = 0.22$; CHCl_3); IR (neat) ν_{max} : 3400 (br, OH), 1680 (C=O), 1515 (C=O) cm^{-1} ; UV/Vis (CH_3CN): λ_{max} (log ϵ): 242 (3.75), 284 (3.90) nm; MS (ESI): m/z (%): 325 $[\text{M} + \text{Na}]^+$; HRMS-ESI: m/z $[\text{M} + \text{Na}]^+$ calcd for $\text{C}_{16}\text{H}_{14}\text{O}_6\text{Na}$: 325.0689; found: 325.0688; ^1H NMR (400 MHz, CDCl_3): 2.70 (1H, dd, $J = 16.8, 3.0$ Hz, H-3a), 3.06 (1H, dd, $J = 16.8, 13.0$ Hz, H-3b), 5.37 (1H, dd, $J = 13.0, 3.0$ Hz, H-2), 3.90 (3H, s, OMe-3'), 6.51 (1H, s, H-8), 6.87 (1H, d, $J = 8.0$ Hz, H-5'), 7.02 (1H, dd, $J = 8.0, 2.0$ Hz, H-6), 7.21 (1H, d, $J = 2.0$ Hz, H-2'), 7.24 (1H, s, H-5). ^{13}C NMR (100 MHz, CDCl_3): δ 44.7 (C-3), 56.2 (3-OCH₃), 80.9 (C-2), 104.3 (C-8), 108.2 (C-5), 112.9 (C-2'), 113.5 (C-10), 115.3 (C-5'), 120.2 (C-6'), 131.2 (C-1'), 144.6 (C-6), 147.2 (C-4'), 148.5 (C-3'), 155.1 (C-7), 159.1 (C-9), 190.5 (C-4, C=O).

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